



# *Electronics Devices (19EC31)*

## *Class 3*

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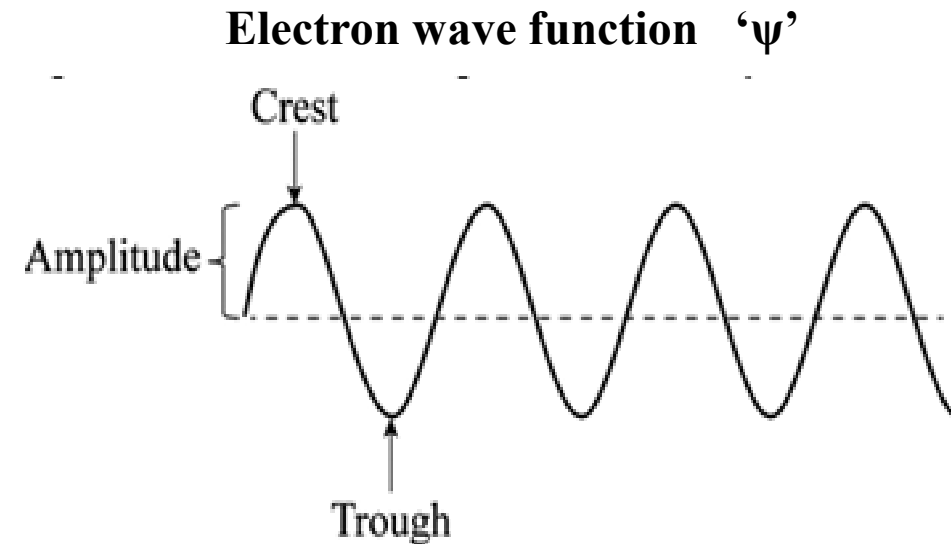
# Electron Wave Function

Electrons in atoms

- 1) Wave (Electron wave)
- 2) Particle

**Electron wave function 'ψ'**

LCAO – Linear Combinations of the individual atomic orbital

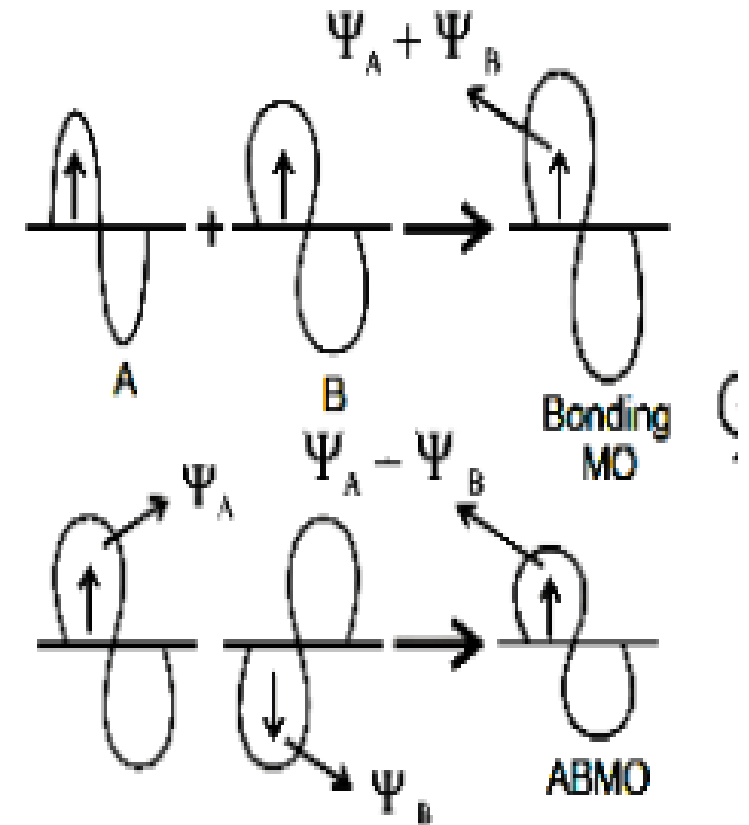
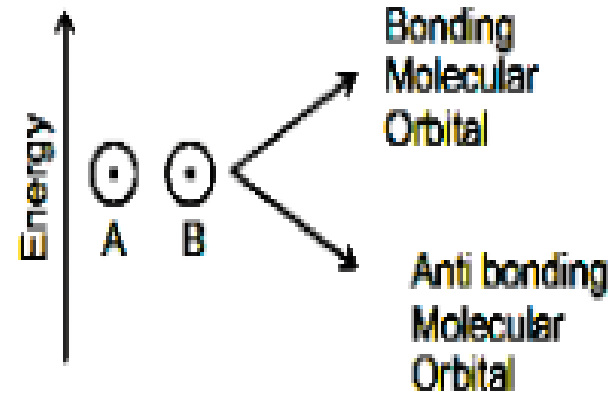


# Molecular Orbits

## Molecular Orbits

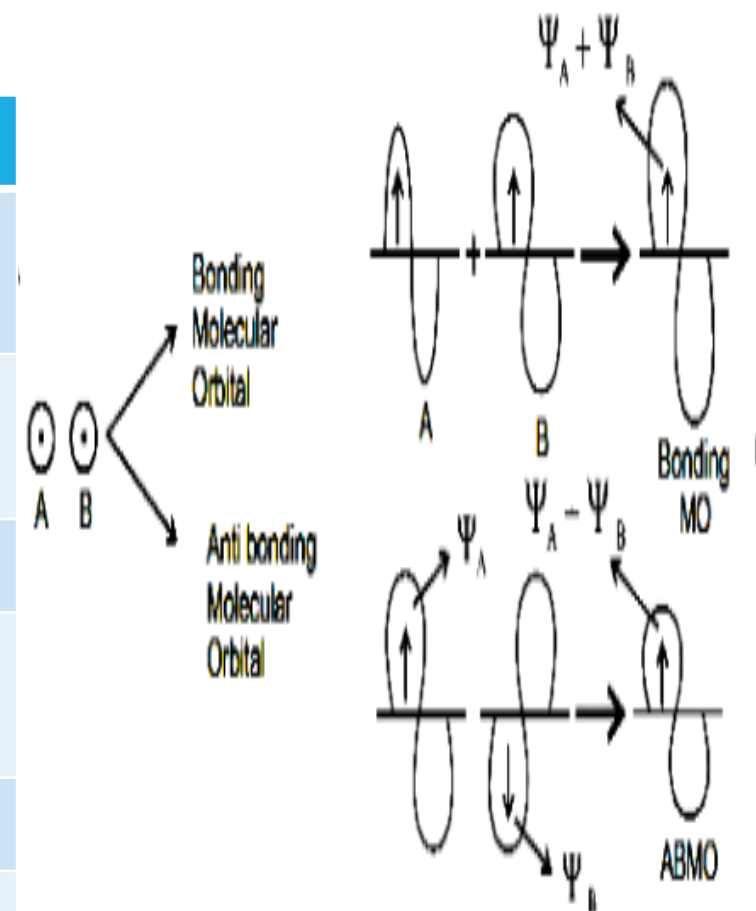
1. Bonding Molecular Orbit (BMO)
2. Antibonding Molecular Orbit (AMBO)

' $\psi_A$ ', ' $\psi_B$ ' ----  
two electron wave  
function



# Molecular Orbits

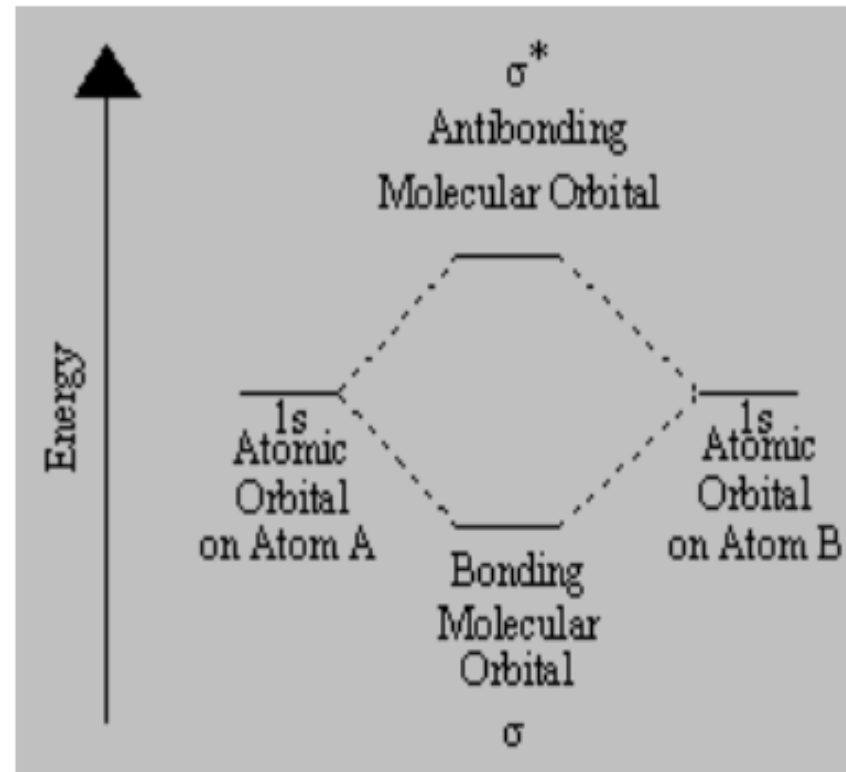
Bonding Molecular Orbit	Antibonding Molecular Orbit
1. $\Psi_{BMO} = \Psi_A + \Psi_B$	$\Psi_{ABMO} = \Psi_A - \Psi_B$
2. Waves are In-Phase Combination	Waves are Out of Phase Combination.
3. Stable Bonding	Unstable Bonding
4. Electron density is more (additive)	Electron density is less (destructive)
5. Constructive Interference	Destructive Interference
6. Occupies Lower energy level because of electron attraction is less	Occupies Higher energy level because of electron attraction is more



# Molecular Orbits

## Molecular Orbits

1. Bonding Molecular Orbit
2. Antibonding Molecular Orbit



higher,  
less stable

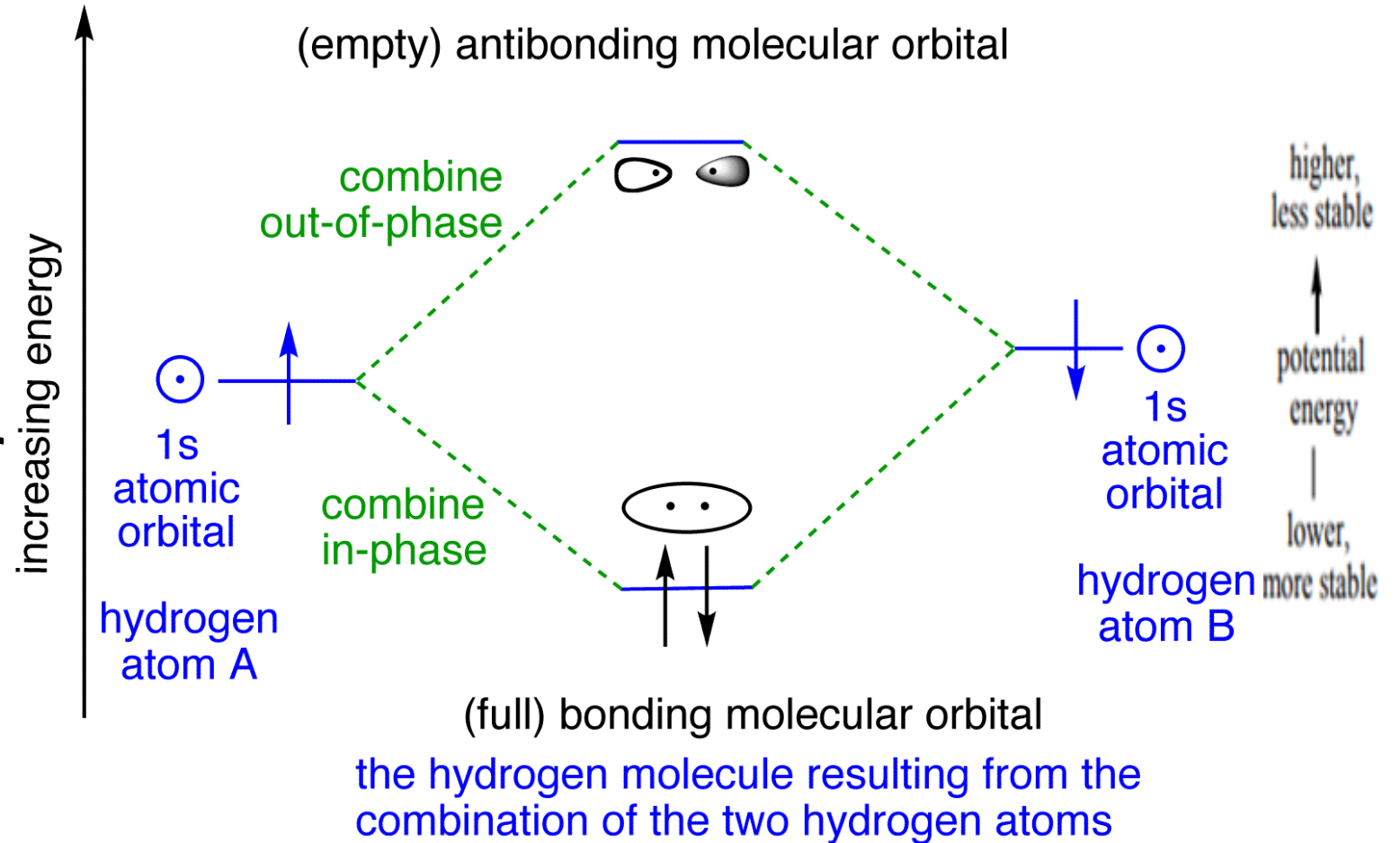
↑  
potential  
energy

↓  
lower,  
more stable

# Molecular Orbits

## Molecular Orbits

1. Bonding Molecular Orbit
2. Antibonding Molecular Orbit





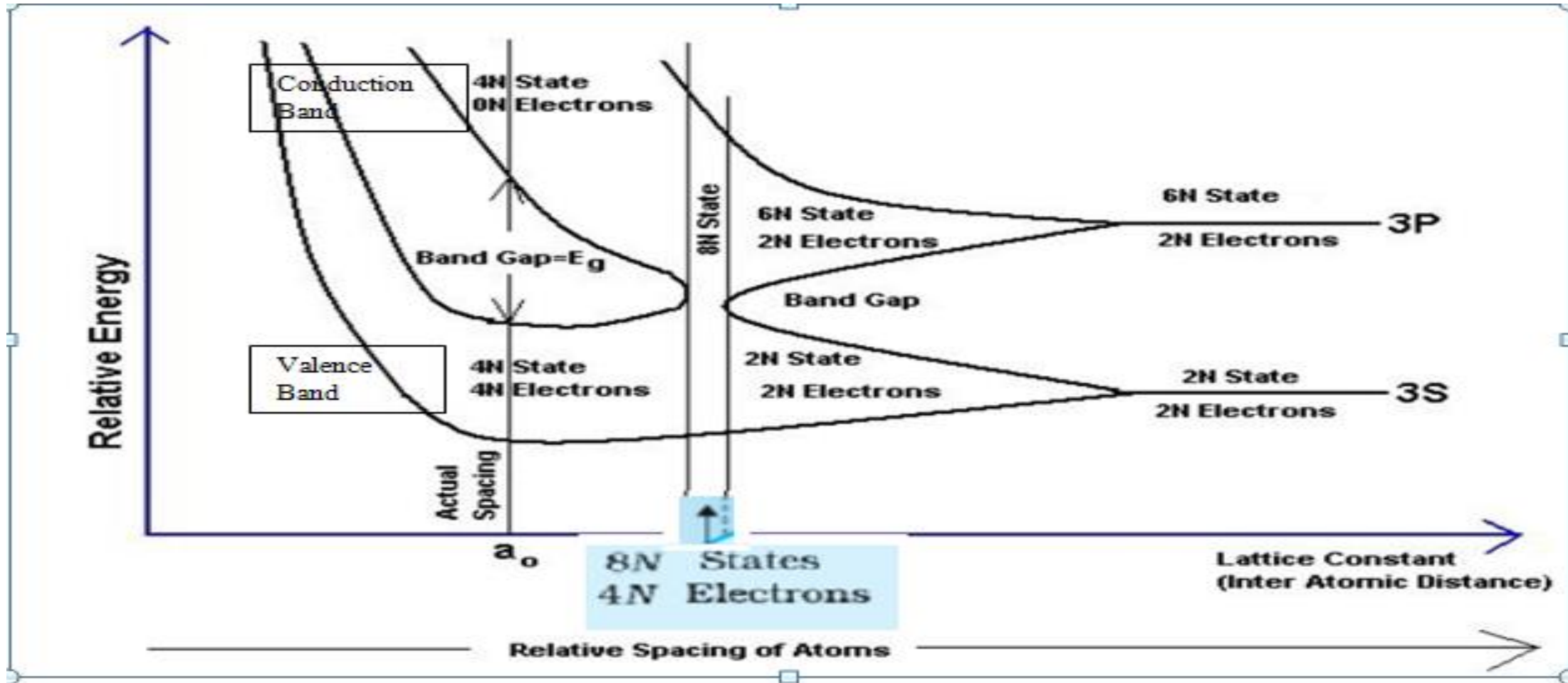
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# *Energy Band Diagram Formation of Silicon Semiconductor*

*Isolated Silicon Atom* -- It means atom which exist separately without any bonding with other atoms.

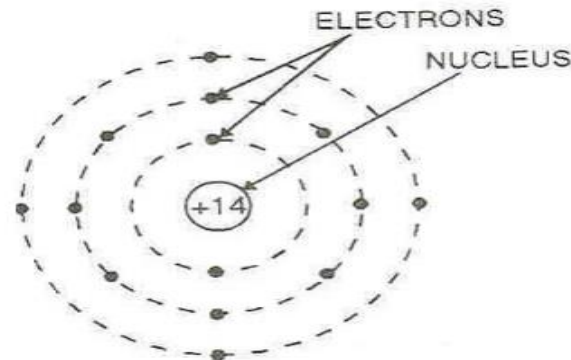
# Energy Band Diagram Formation





# *Energy Band Diagram Formation of Silicon Semiconductor*

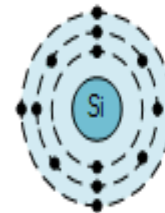
- Silicon Atom - 14
- Si –  $1s^2 2s^2 2p^6 3s^2 3p^2$



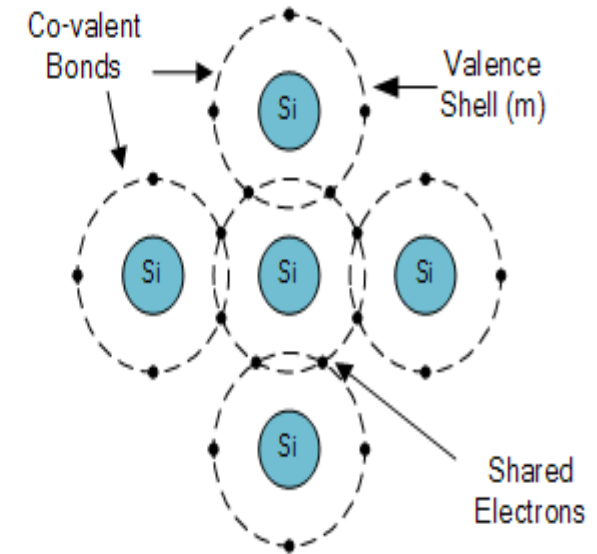
**Atomic structure of Silicon**

Valence band of Si is  $3s^2 3p^2$   
 Valence electrons of Si = 4

A Silicon Atom,  
Atomic number = "14"

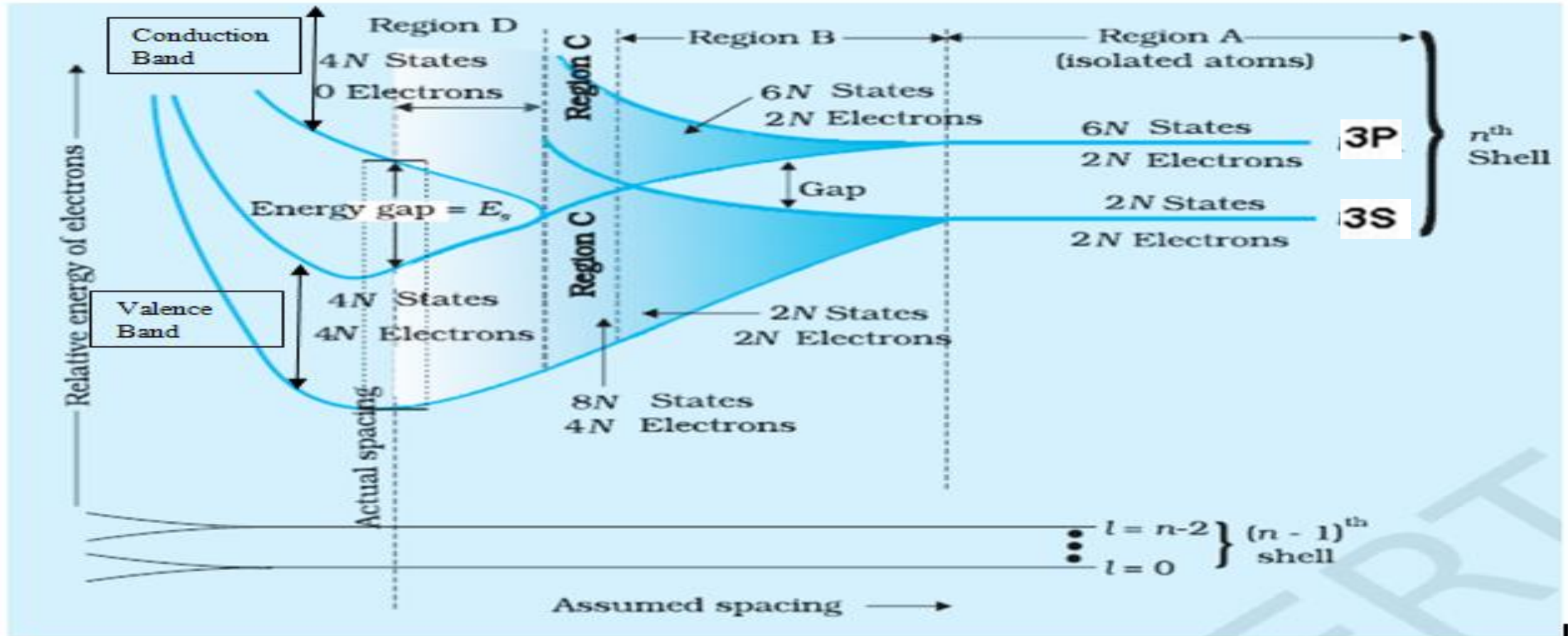


Silicon atom showing 4 electrons in its outer valence shell (m)

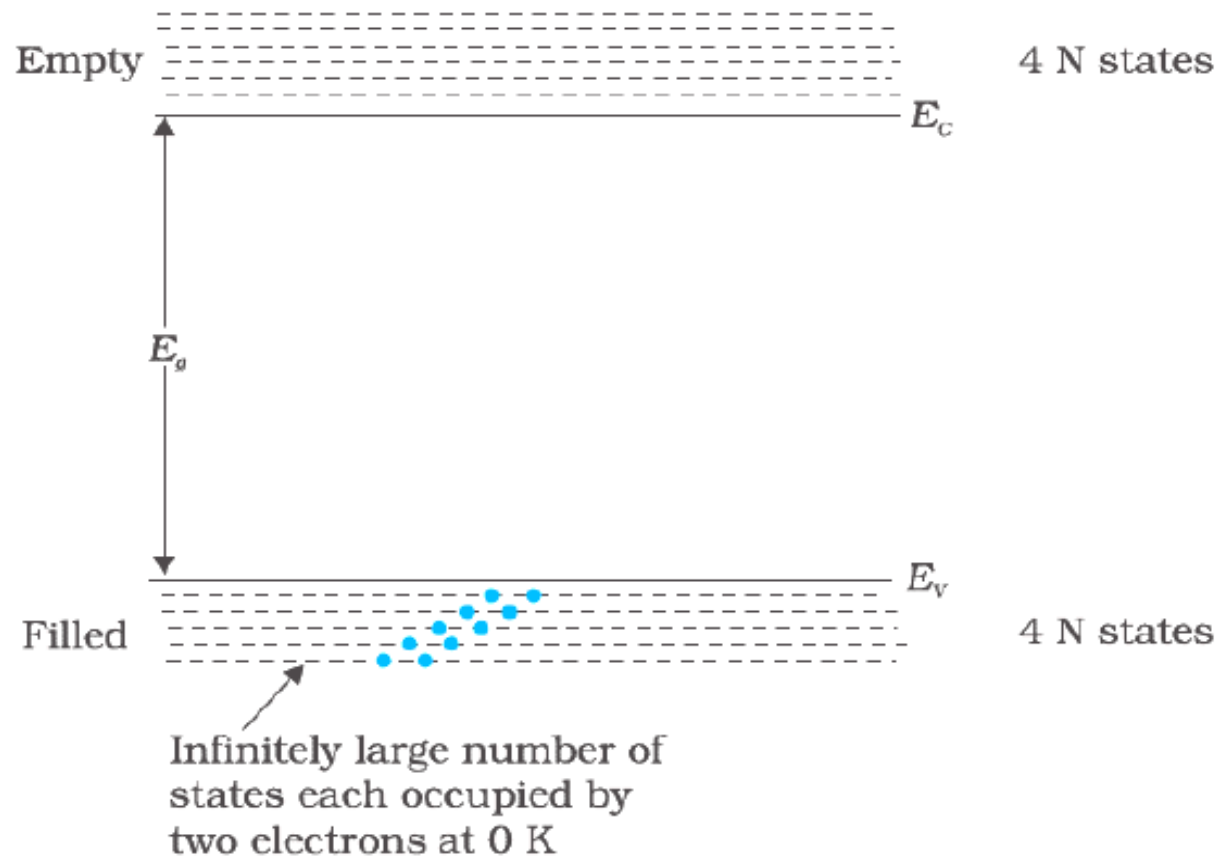


Silicon Crystal Lattice

## Energy Band Diagram Formation



## Energy Band Diagram Formation



- **At 0 K**, The lower band , called the valence band, having  $4N$  energy states, completely filled with  **$4N$  electrons**
- The upper band , called the Conduction band having  $4N$  energy states with completely **ZERO electrons**
- **At absolute zero (0K), Silicon acts like a perfect Insulator**



Dr. Ambedkar  
Institute of Technology



*Thank You*

*HAVE A NICE DAY*